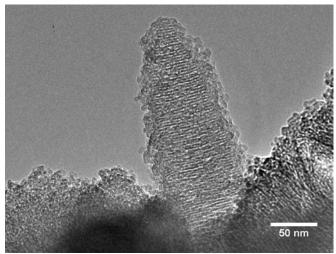
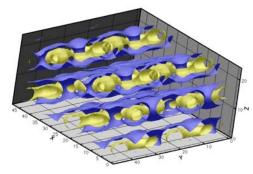
## NIRT: Tailored Fluorinated Surfactants for the Design of Ordered Nanoporous Ceramics

S.E. Rankin and B.L. Knutson, University of Kentucky H.J. Lehmler, University of Iowa, DMR-0210517

We combine chemical synthesis, materials synthesis and Monte Carlo simulations to develop ceramics with well-defined nanoscale pores by spontaneous assembly of ceramic precursors and surfactants. The fluorinated surfactants that we use are "non-stick" molecules (like Teflon®) and give smaller pores, new pore arrangements, and better processing strategies. In one application (lower left), we prepared functional particles that hold and protect a pink dye in an acidic environment, but release it when neutralized Recent results have been published in Langmuir and Microporous and Mesoporous Materials.







Top: Transmission electron micrograph of elongated fluorinated surfactant-templated particles
Left: pH responsive particles loaded with colored dye
Right: Computer simulation of pore structure

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## **Outreach:**

- •Meetings with local industry participation:
  - •Tristate Catalysis Society
  - •Kentucky Nanomaterials Workshops
  - •Innovation and Enterprise Conferences
- •E-day Open House (2,000 K-12 students)
  - •"Materials World" nanotechnology poster
  - •Hands-on separation using porous particles
- •Engineering camp for high school juniors
- •Feature\* in *ODYSSEY* magazine for local research community



## **Education:**

- •In addition to graduate students and postdocs, 8 undergraduate students have participated in project
- •One REU student
- •Two RCTF Minority Scholars
- •One undergraduate student co-authored a publication in print
- •Two undergraduate students presented results at American Institute of Engineers National Meeting, Fall 2003
- •One alumnus from project went on to become an NSF Graduate Fellow at University of Minnesota

\*http://www.rgs.uky.edu/ca/odyssey/spring03/underpressure.html

■ Tave you ever noticed those little paper packets hiding under the wads of crumpled tissue paper in a box of new shoe? You know, the ones that read: Silica gel. Throw Away, DO NOT EAT. As a kid. I always wondered what would happen if I tore off a corner and popped one of those spheres into my mouth. Now II know.

P. Gregory Silica gel is hydrophilic—it likes water. Through the science of adsorption, the process by which molecules of a liquid or gas contact and